

# In vitro study of the antimicrobial activity of European propolis against *Paenibacillus larvae*

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**Abstract** – Typical European propolis is an antibiotic substance derived from the resin of poplar, birch, and aspen buds. Here, we tested the ability of propolis originating from the resins of these plants to inhibit the in vitro growth of *Paenibacillus larvae*, the organism that causes American foulbrood, a fatal honeybee larval disease. The study involved GC-MS analysis of extracts from nine samples of propolis gathered from the temperate climate zone of Europe. The extracts showed noticeable differences in the content of flavonoids and other phenolic compounds. Despite the differences in chemical composition, all tested extracts inhibited the growth of *P. larvae*, with a minimum inhibitory concentration of 7.8 to 62.4  $\mu\text{g mL}^{-1}$ . It was found that not only did the content of flavonoids affect the strength of antimicrobial activity, but other phenolics, such as the phenylpropanoids hydroxycinnamyl sesquiterpenols, glycerides and benzoates also had an effect. This is the first report on the comparative activity of different types of European propolis against *P. larvae*.

*Paenibacillus larvae* / American foulbrood / *Apis mellifera* / propolis / antibacterial action

## 1. INTRODUCTION

American foulbrood (AFB), a disease of honeybee larvae, has been known for more than 200 years. Prior to the arrival into Europe and North America of the parasitic mite *Varroa destructor*, AFB was the most economically important disease of honeybees. To date, AFB is the most infectious and destructive disease of honeybee brood and is often fatal for bee colonies (Genersch 2010). In Poland, AFB is the only notifiable disease of honeybees and is subject to obligatory combating (Buczek 2011). The final

decision regarding the method of combating (treating or destroying AFB colonies) is within the purview of the district veterinary surgeon.

The causative agent of this disease is the Gram-positive, spore-forming bacterium *Paenibacillus larvae*, which is highly infectious. In some countries, antibiotics or sulfathiazole are used to combat AFB; however, these drugs only suppress the clinical symptoms and cannot cure the disease, because they are not effective against the infectious endospores (Genersch 2010). Furthermore, this practice leads to serious consequences, such as reduced honeybee vitality through the suppression of the endogenous microflora in these bees, the almost inevitable pollution of beekeeping products with medicinal residues, and the appearance of bacterial resistance (Lodesani and Costa 2005; Miyagi et al. 2000; Gende et al. 2010; Buczek 2011). Antibiotics and sulfathiazole are

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legally banned in the European Union for use in beekeeping (Mutinelli 2003), because honeybees are classed as farming animals, whose products cannot contain the residues of any drugs; therefore, this greatly limits the treatment possibilities.

Practically the only nonmedicinal method that is applied to sanitize infected colonies is the so-called shook swarm method (proposed in 1769 by Schirach), which consists of shaking adult bees onto a new uninfected comb foundation and destroying the old combs and other infected materials (Hansen and Brødsgaard 2003). This method can lead to positive results in the treatment of infected but not clinically diseased colonies (Genersch 2010). However, its application does not guarantee the complete recovery of treated bee colonies. Therefore, in recent decades, considerable attention has been paid to the development of alternative treatment methods, based on the application of natural antibacterial substances such as essential oils (Fuselli et al. 2008; Gende et al. 2008, 2009, 2010; González and Marioli 2010; Santos et al. 2012) or propolis (Antúnez et al. 2008; Bastos et al. 2008; Mihai et al. 2012; Bilikova et al. 2013; Wilson et al. 2015).

Propolis is a mixture of beeswax and resinous material that is collected by honeybees from various plants. In the mid-latitudes of the northern hemisphere, the plant precursor of propolis is the bud resins (exudates) of some arboreous trees, in particular, different species of poplar, aspen, and birch (Bankova et al. 2000; Wilson et al. 2013, 2015; Isidorov et al. 2014a, 2016). The antimicrobial activity of this natural antibiotic has been attributed to phenolic substances: flavonoid aglycones, and phenolic and hydroxycinnamic acids (Kujumgiev et al. 1999). Their specific content tends to vary to a great extent, depending on the plant precursor (Bankova et al. 2006).

Recent publications have demonstrated that the principal plant precursors of propolis from boreal and temperate zones of the European continent are the bud resins of black poplar (*Populus nigra*), downy birch (*Betula pubescens*), and common aspen (*Populus tremula*) (Isidorov et al. 2014a). In a more recent investigation, taxonomical markers of the resins from the buds of these trees were determined (Isidorov et al. 2016). Black poplar bud resin is characterized by a high content of phenols such as pentenyl (mostly, prenyl)

cinnamates, chalcones, and the unsubstituted B ring flavonoids: pinocembrin, pinostrobin, chrysin, galangin, pinobanksin, and their 3-substituted derivatives (Isidorov and Vinogorova 2003). Downy birch and common aspen bud resins are distinguished by the presence of specific phenylpropanoids, hydroxycinnamic acid esters of sesquiterpene alcohols, and glycerol, respectively (Isidorov et al. 2014a, 2016). Although bees from the same colony often collect resins from more than one of these tree species, three main “types” of European propolis can be distinguished on the basis of the species-specific chemical composition: poplar-, birch-, and aspen-type (Popova et al. 2012; Isidorov et al. 2014a, 2016). Probably, the differences in the chemical composition are reflected in the antimicrobial activity of different “types” of propolis; however, little is known about the effects of different types of European propolis on *P. larvae*.

The main aim of this communication was to compare the *in vitro* anti-*P. larvae* action of three types of propolis from the temperate zone of the northern hemisphere in an attempt to relate the observed minimal inhibition concentration (MIC) values with the chemical composition of these bee products.

## 2. MATERIALS AND METHODS

### 2.1. Chemicals and propolis

Pyridine, bis(trimethylsilyl)trifluoroacetamide (BSTFA), and C<sub>8</sub>-C<sub>40</sub> *n*-alkane calibration standards were purchased from Sigma-Aldrich, (Poznań, Poland). Extractions were carried out with diethyl ether (POCH SA, Gliwice, Poland).

Two propolis samples (Pr-1 and Pr-2) were collected in the same apiary located in northeastern Poland, on the outskirts of Białowieza National Park (52° 42' N, 23° 52' E). Propolis Pr-3 originated from the forest-steppe in the Volgograd region, Russia (48° 48' N, 44° 28' E). Sample Pr-4 was supplied from the Slovak Republic (48° 44' N, 19° 08' E). Three Russian samples originated from the taiga zone: propolis Pr-5 was collected in the Perm' region (57° 40' N, 56° 12' E), while Pr-6 and Pr-7 samples were received from apiaries in the Vologda region

(59° 58' N, 38° 31' E and 59° 47' N, 38° 38' E). Sample Pr-8 originated from Finland (61° 43' N, 25° 26' E), and sample Pr-9 was supplied from northeastern Latvia (57° 8.5' N, 26° 27' E).

Propolises Pr-1, Pr-2, and Pr-9 were collected by the authors in the summer of 2015. To acquire the material, a special net (mesh size of 1 mm) was mounted just above the hive frames with the brood. After a time interval of 3 weeks, the net became fully glued with pure propolis by the bees. The propolis was easily separated after cooling the net to  $-18^{\circ}\text{C}$ . The remaining samples (10–15 g each) were gathered in the summer of 2015 by apiarists in different countries. Propolis was harvested by scraping it off the frames or by using a plastic screens which were placed on the topmost frames in the hive and leaved until the bees have deposited propolis in the splits in the screen.

## 2.2. Extract preparation and analyses

Two grams of ground propolis powder was transferred to a 100 mL retort and extracted with three 50 mL portions of diethyl ether for 24 h. The joint extracts were filtered through paper filter, and the solvent was removed using a rotary evaporator. The dry residue was used for chemical analysis and antibacterial tests.

About 5 mg of the residue was diluted with 220  $\mu\text{L}$  of pyridine and 80  $\mu\text{L}$  of BSTFA. The mixture was sealed and heated for 30 min at  $60^{\circ}\text{C}$  to form trimethylsilyl (TMS) derivatives. TMS derivatives were analyzed using GC-MS on an HP 6890 gas chromatograph fitted with an MSD 5973 mass selective detector (electron impact source and quadrupole analyzer) from Agilent Technologies (USA). This device was equipped with an HP-5MS fused silica column (30 m  $\times$  0.25 mm i.d., 0.25- $\mu\text{m}$  film thickness) with electronic pressure control and a split/splitless injector. The latter was used at  $250^{\circ}\text{C}$  in split (1:50) mode. The helium flow rate through the column was 1 mL  $\text{min}^{-1}$  in constant flow mode. The column was charged with 1  $\mu\text{L}$  of the liquid sample containing about 0.3  $\mu\text{g}$  of derivatized components. The analysis was carried out with temperature programming from 50 to  $310^{\circ}\text{C}$  at a rate of  $5^{\circ}\text{C min}^{-1}$ ; the highest temperature was maintained for 15 min. The MSD detector

acquisition parameters were as follows: the transfer line temperature was  $280^{\circ}\text{C}$ , the MS source temperature was  $230^{\circ}\text{C}$ , and the MS quad temperature was  $150^{\circ}\text{C}$ . The electron impact mass spectra were obtained at an ionization energy of 70 eV. The MSD was set to scan 41–600 amu.

After integration of the chromatogram, the fraction of each component in the total ion current (TIC) was calculated. The precision of the method was studied by three replicate extractions and analyses. The peak areas of the extract components obtained by replicate analyses were used for the calculation of their relative standard deviation (RSD) values. On average, RSD amounted to 2% for the main peaks (more than 10% of TIC), 6% for medium peaks (more than 1% of TIC), and 18% for peaks that amounted to  $\leq 0.5\%$  of TIC (Isidorov et al. 2016).

To identify the components, both mass spectral data and the calculated retention indices were used. Mass spectrometric identification was carried out with an automatic system of GC-MS data processing supplied by the NIST 14 library (NIST/EPA/NIH Library of Electron Ionization Mass Spectra containing 276,259 standard mass spectra) and a home-made library of mass spectra. The latter contains more than 1250 spectra of TMS derivatives prepared from commercial preparations of flavonoids, other phenolics, terpenoids, aliphatic and aromatic acids, and alcohols.

The linear temperature programmed retention indices ( $I^T$ ) were calculated from the results of the separation of  $\text{C}_8$ – $\text{C}_{40}$   $n$ -alkane solutions in hexane and the TMS derivatives. The measured values of the retention times were used to calculate the retention indices using the following equation:

$$I^T = 100 (t_x - t_n) / (t_{n+1} - t_n) + 100n$$

where  $t_x$ ,  $t_n$ , and  $t_{n+1}$  are the retention times of compound  $x$  and  $n$ -alkanes with the number of carbon atoms in the molecule  $n$  and  $n+1$ , respectively ( $t_n \leq t_x \leq t_{n+1}$ ) (Van den Dool and Kratz 1963).

The obtained  $I^T$  values were compared with the NIST collection (NIST 2013) as well as with the authors' previously published data (Isidorov et al. 2009, 2014a, b; Isidorov 2015) and presented in a home-made computer database containing more than 18,000  $I^T$  values for more than 5500 organic

compounds. The identification was considered reliable if the results of the computer search of the mass spectra library were confirmed by the experimental  $I^T$  values, i.e., if their deviation from the home-made database values did not exceed  $\pm 5$  u.i. (the highest quantity of intra-laboratorial deviation).

### 2.3. *P. larvae* isolates

*P. larvae* was isolated from honeybee larvae and honey samples originating from apiaries in southeastern Poland (Table I). To isolate the bacilli, dead larvae were aseptically removed from brood combs and were crushed and suspended in 5 mL of physiological saline (0.9% NaCl), followed by shaking for 10 min at room temperature. To isolate bacteria from honey samples, about 10 mL of honey was preheated at 45 °C, diluted with sterile water in a ratio of 1:1, and centrifuged at 3000×g for 30 min. The resulting pellets were suspended in 200 µL of physiological saline. The solutions of crushed larvae and honey were centrifuged at 10,000 rcf for 5 min, and the resulting pellet was then suspended in 900 µL of physiological saline. All samples were preheated in a water bath for 10 min at 85 °C to eliminate vegetative cells and to select the endospores. After preparation of a serial dilution ( $10^{-1}$ – $10^{-4}$ ), 100 µL of each solution was spread onto Columbia Blood Agar Base supplemented with vitamin B<sub>1</sub> (Oxoid, Basingstoke, England) and incubated at 30 °C in 5% CO<sub>2</sub> for 48–96 h. The bacteria that formed round colonies with a diameter of about 2–3 mm and that were transparent or slightly whitish, with an elevated center and frayed edge, were initially selected as *P. larvae* and were cultivated on the same medium and under the same conditions mentioned above to obtain pure cultures. The bacilli

initially classified as *P. larvae* were further investigated by Gram staining, the catalase test (Neuendorf et al. 2004), and a Plagemann probe. Gram-positive and catalase-negative bacilli that formed spiral forms in the Plagemann probe were classified as *P. larvae* (Plagemann 1985). The selected strains were kept at –80 °C in Mueller-Hinton broth (Oxoid) supplemented with glycerol at a ratio of 1:1.

### 2.4. 16S rRNA gene sequencing

Total DNA was prepared from overnight cultures of the isolates grown in a brain-heart infusion (BHI) broth using the protocol for Gram-positive bacteria with the DNeasy Blood & Tissue Kit (Qiagen GmbH, Hilden, Germany) and the Qiacube apparatus (Qiagen), according to the instruction manuals. The DNA concentration and purity were checked using a NanoDrop 2000 spectrometer (Thermo Fisher Scientific Com., Waltham, USA). The partial 16S rRNA gene of the isolates was amplified using a pair of primers: PI-16SE1 5'-GCA AGT CGA GCG GAC CTT GTG-3' and PI-16SE2 5'-AAA CCG GTC AGA GGG ATG TCA AG-3' (Neuendorf et al. 2004) in a PCR reaction using the Expand Long Template PCR System (Roche Diagnostic GmbH, Mannheim, Germany) as follows: 94 °C for 3 min, 94 °C for 30 s, 50 °C for 45 s, and 68 °C for 7 min. The 966-bp 16S rRNA amplicons were cloned into pGEM-T Easy (Promega Corporation, Madison, USA). To determine the accuracy of the sequence, both strands of two clones were sequenced using T7 and SP6 primers, in an ABI3500 automated sequencer (Applied Biosystems, Foster City, USA). For the comparative analyses of nucleotide and amino acid sequences, database searches were performed using the BLAST program at the NCBI website (<http://www.ncbi.nlm.nih.gov>). This study confirmed the identification of isolates KB25, KB35, KB41, and KB55 as *P. larvae*.

### 2.5. Anti-*P. larvae* susceptibility test

The propolis extracts were tested against four “wild” *P. larvae* isolates, as well as against the reference *P. larvae* LMG 09820 strain (Belgian

**Table I.** *Paenibacillus larvae* isolates used in the study.

Name of isolate	Origin	Country	Region
KB25	Larvae	Poland	Lubelskie
KB35	Larvae	Poland	Lubelskie
KB41	Honey	Poland	Podkarpackie
KB55	Larvae	Poland	Podkarpackie

Coordinated Collections of Microorganisms). The latter was isolated in 1906 from a foul brood of honeybees by White as *Bacillus larvae* (White 1906); this was reclassified several times and was finally named *Paenibacillus larvae* (Heyndrickx et al. 1996). The bacterial strains kept at  $-80\text{ }^{\circ}\text{C}$  were inoculated onto blood agar (Oxoid) and were incubated for 48 h at  $30\text{ }^{\circ}\text{C}$ . Then, bacteria were reinoculated onto Mueller-Hinton broth and cultured until they reached an optical density of 0.2–0.4 at 600 nm, measured with a V-670 spectrophotometer (Jasco, Japan).

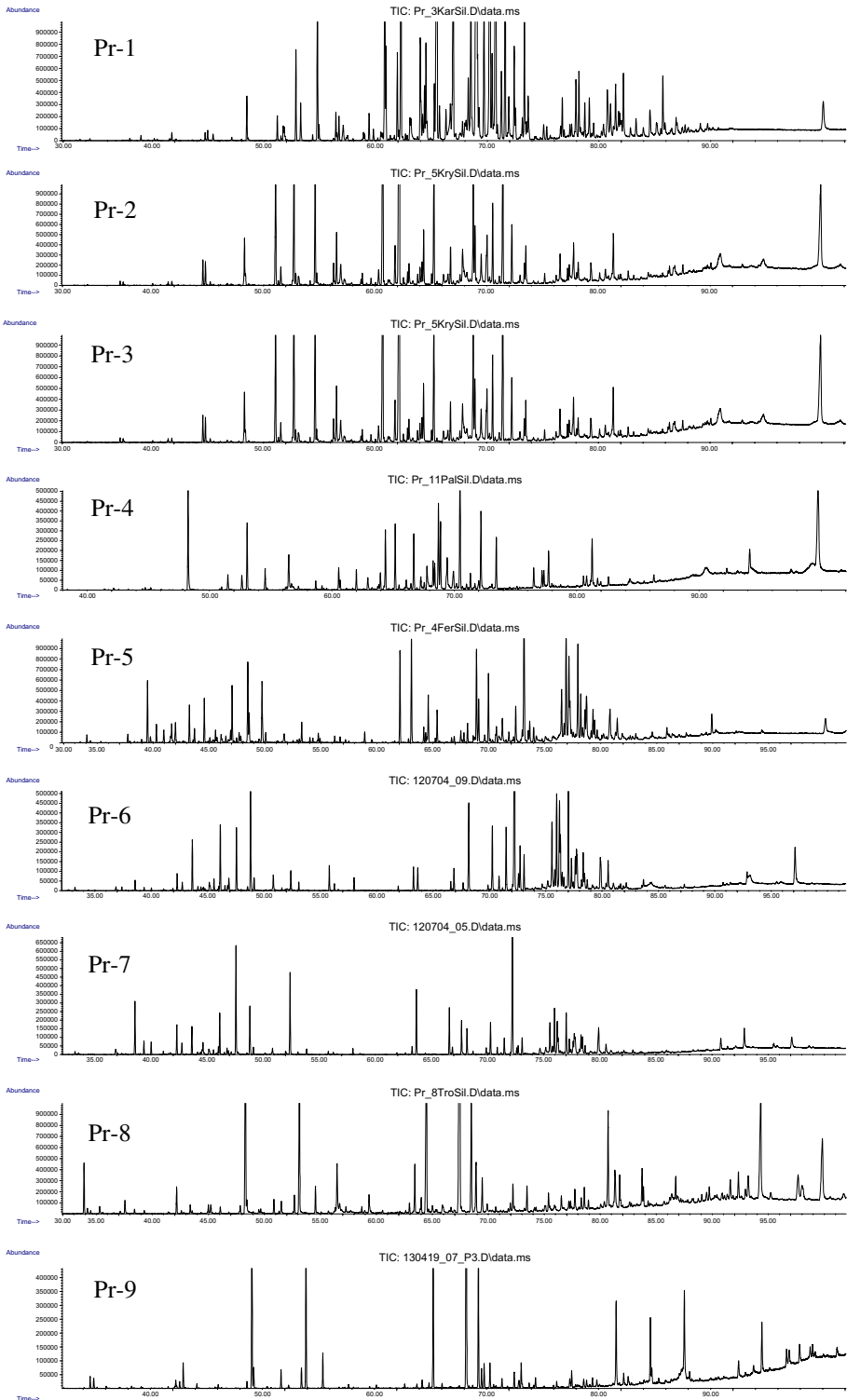
The propolis extracts were dissolved in DMSO at a concentration of  $2\text{ mg mL}^{-1}$ , sonicated in a Branson 2510 ultrasonic bath (Sigma), and filtered through a 0.22- $\mu\text{m}$  pore size Rotilabo-Spritzenfilter filter (Carl Roth GmbH and Co, Karlsruhe, Germany). To maintain appropriate nutrient conditions, the extracts were aseptically diluted in double-concentrated Mueller-Hinton broth using laboratory tubes that contained an arithmetic dilution series ranging from 0.5 to  $1000\text{ }\mu\text{g mL}^{-1}$  in a final volume of 3 mL. The propolis solutions used in the study were as follows: 0.5, 1.0, 2.0, 3.9, 7.8, 15.6, 31.2, 62.5, 125.0, 250, 500, and  $1000\text{ }\mu\text{g mL}^{-1}$ . In order to observe whether the solvent alone caused turbidity of the medium, we included a solvent control.

The minimal inhibitory concentration (MIC) of the extracts was assessed in the tube dilution test in accordance with Clinical and Laboratory Standard Institute (2011) protocols. For the assay, 30  $\mu\text{L}$  of the bacterial culture of an  $\text{OD}_{600} \sim 0.4$  was added to each tube with the extract and was incubated for 48 h at  $37\text{ }^{\circ}\text{C}$ . The bacterial cultures with different concentrations of propolis were observed visually. The lowest dose of the extract with no growth of *P. larvae* was regarded as the MIC value. All the tests were repeated four times. Cultures of *P. larvae* in tubes with Mueller-Hinton broth without propolis extracts were used as a positive control. Mueller-Hinton broth supplemented with 10% DMSO arithmetically diluted ranging from 0.0025 to 5% was used as a solvent control. Similar growth of bacteria under study in Mueller-Hinton broth without DMSO (positive control) and in Mueller-Hinton broth supplemented with 10% DMSO (solvent control) was observed, which allowed to conclude that DMSO itself does not affected *P. larvae* growth.

### 3. RESULTS

The composition of diethyl ether extracts of propolis was very complex: on the chromatograms of nine samples of propolis from five European countries (Figure 1), 278 organic components were recorded, 141 of which belonged to different groups of aromatic compounds. More comprehensive groups were formed by flavonoids and chalcones (53 unique chromatographic peaks) and by phenylpropanoids (63 unique peaks for cinnamic acid derivatives). Apart from hydroxycinnamic acids and their benzyl and pentenyl esters (36 peaks), phenylpropanoids were also represented by the hydroxycinnamoyl esters of sesquiterpene  $\text{C}_{15}\text{H}_{24}\text{O}$  alcohols (10 peaks), as well as monoglycerides and diglycerides of *p*-coumaric, ferulic, and caffeic acids (17 peaks). Furthermore, all propolis samples contained several terpenoids in different amounts: the chromatograms showed the presence of  $\text{C}_{15}\text{H}_{24}\text{O}$ ,  $\text{C}_{15}\text{H}_{26}\text{O}$ , and  $\text{C}_{15}\text{H}_{24}\text{O}_2$  sesquiterpenoids (39 peaks), and  $\text{C}_{30}\text{H}_{48}\text{O}$ ,  $\text{C}_{30}\text{H}_{50}\text{O}$ , and  $\text{C}_{30}\text{H}_{50}\text{O}_2$  triterpenoids (19 peaks). In general, the qualitative composition of the samples under investigation corresponded well with previously published data on European propolis (Bankova et al. 2000, 2006; Isidorov et al. 2014a). Table II presents the group composition of extracts in terms of their proportion of the total ion current (TIC) of the chromatogram.

The specific features of the qualitative composition of the propolis samples under investigation led to the conclusion that among the nine samples listed in Table II, only three, i.e., Pr-1, Pr-5, and Pr-9, were selectively collected by honeybees from a single plant source. The propolis Pr-1 was almost a pure poplar-type propolis, Pr-5 was a birch type, and Pr-9 was an aspen-type propolis. Samples Pr-2, Pr-3, and Pr-4 also belonged to the poplar-type; however, they contained a relatively small admixture (0.6–1.0% of TIC) of phenylpropanoid glycerides, which are taxonomical markers of aspen bud resin (Isidorov et al. 2009). Samples Pr-6 and Pr-7 of Russian propolis were birch-type, but contained marked amounts (2.4 and 7.8% of TIC) of aspen resin-derived phenylpropanoid glycerides. Aspen-type propolis Pr-8 contained an admixture (2.5% of TIC) of typical downy birch phenylpropanoid sesquiterpenols.



**Figure 1.** GC-MS chromatograms of diethyl extracts of propolis samples Pr-1 to Pr-9.

**Table II.** Relative group composition (% of TIC,  $n = 3$ ) of different types of European propolis.

Group of compounds	Polish, Pr-1	Polish, Pr-2	Russian, Pr-3	Slovak, Pr-4	Russian, Pr-5	Russian, Pr-6	Russian, Pr-7	Finnish, Pr-8	Latvian, Pr-9
Flavonones, including	<b>13.3 ± 0.8</b>	<b>4.9 ± 0.1</b>	<b>8.4 ± 0.5</b>	<b>8.0 ± 0.5</b>	<b>15.3 ± 0.9</b>	<b>12.1 ± 0.7</b>	<b>4.0 ± 0.1</b>	<b>1.0 ± 0.2</b>	<b>0.2 ± 0.2</b>
Pinocembrin	11.4 ± 0.2	3.1 ± 0.2	4.7 ± 0.3	5.3 ± 0.3	-	-	-	-	-
Pinostrobin	1.7 ± 0.3	1.4 ± 0.2	0.8 ± 0.1	0.3 ± 0.1	-	-	-	-	-
Sakuranetin	0.2 ± 0.2	0.2 ± 0.2	0.2 ± 0.2	0.2 ± 0.2	9.2 ± 0.5	6.5 ± 0.4	2.6 ± 0.2	0.8 ± 0.1	0.1 ± 0.1
Homoeriodictyol	-	-	-	0.5 ± 0.2	4.1 ± 0.2	2.4 ± 0.4	0.8 ± 0.2	-	-
Flavones, including	<b>9.0 ± 0.2</b>	<b>5.8 ± 0.1</b>	<b>6.2 ± 0.4</b>	<b>9.0 ± 0.5</b>	<b>13.8 ± 0.3</b>	<b>6.0 ± 0.4</b>	<b>2.9 ± 0.5</b>	-	-
Chrysin	8.4 ± 0.2	4.9 ± 0.1	4.0 ± 0.2	6.8 ± 0.1	-	-	-	-	-
4'-7-Dimethyl apigenin	-	-	0.7 ± 0.1	0.9 ± 0.2	2.4 ± 0.1	1.2 ± 0.01	0.5 ± 0.1	-	-
4',7-Dimethyl kaempferol	-	-	0.6 ± 0.1	0.5 ± 0.1	4.1 ± 0.2	2.1 ± 0.1	0.8 ± 0.1	-	-
Pectolinarigenin	-	-	0.6 ± 0.1	0.5 ± 0.1	5.0 ± 0.1	1.0 ± 0.2	0.9 ± 0.2	-	-
Flavonols, including	<b>12.4 ± 0.7</b>	<b>10.1 ± 0.6</b>	<b>9.6 ± 0.7</b>	<b>15.1 ± 0.3</b>	<b>7.6 ± 0.5</b>	<b>8.8 ± 0.02</b>	<b>3.5 ± 0.1</b>	-	-
Galangin	8.7 ± 0.2	7.2 ± 0.4	4.9 ± 0.3	9.1 ± 0.2	-	-	-	-	-
4-Methyl kaempferol	-	-	0.2 ± 0.1	0.2 ± 0.2	2.2 ± 0.4	4.8 ± 0.3	2.4 ± 0.2	0.2 ± 0.2	-
Dihydroflavonols, including	<b>25.2 ± 1.2</b>	<b>15.3 ± 0.3</b>	<b>8.0 ± 0.5</b>	<b>16.2 ± 0.6</b>	-	-	-	-	-
Pinobanksin	4.2 ± 0.3	2.9 ± 0.2	2.0 ± 0.2	3.7 ± 0.2	-	-	-	-	-
Pinobanksin-3-acetate	17.4 ± 0.3	8.5 ± 0.2	5.5 ± 0.3	11.6 ± 0.2	-	-	-	-	-
Chalcones	<b>1.6 ± 0.1</b>	<b>0.5 ± 0.1</b>	<b>1.0 ± 0.2</b>	<b>1.3 ± 0.3</b>	-	-	-	-	-
Cinnamic acids and esters, including	<b>15.7 ± 0.9</b>	<b>13.5 ± 0.9</b>	<b>14.4 ± 0.8</b>	<b>13.0 ± 0.9</b>	<b>10.8 ± 0.6</b>	<b>16.2 ± 0.4</b>	<b>22.3 ± 0.7</b>	<b>39.4 ± 0.8</b>	<b>36.5 ± 1.5</b>
Hydroxycinnamic acids	3.9 ± 0.7	6.9 ± 0.2	7.4 ± 0.4	5.2 ± 0.5	6.1 ± 0.3	8.3 ± 0.5	10.8 ± 0.6	8.9 ± 0.5	11.9 ± 0.7
Benzyl hydroxycinnamates	4.0 ± 0.3	4.4 ± 0.8	4.6 ± 0.4	3.9 ± 0.5	4.7 ± 0.4	3.7 ± 0.4	11.7 ± 0.5	11.7 ± 0.5	3.8 ± 0.5
Pentenyl hydroxycinnamates	7.8 ± 0.2	2.2 ± 0.6	2.4 ± 0.5	3.1 ± 0.6	-	-	-	-	-
Coniferyl and 4-hydroxycinnamyl benzoates	-	-	-	-	-	-	-	18.7 ± 0.9	20.9 ± 0.4
Phenylpropanoid sesquiterpenols	-	-	-	-	<b>5.5 ± 1.0</b>	<b>4.1 ± 0.7</b>	<b>1.3 ± 0.3</b>	<b>2.5 ± 0.7</b>	-
Phenylpropanoid glycerides	-	<b>0.6 ± 0.2</b>	<b>1.0 ± 0.2</b>	<b>0.8 ± 0.3</b>	<b>0.02 ± 0.02</b>	<b>2.4 ± 0.4</b>	<b>7.8 ± 0.9</b>	<b>30.2 ± 0.6</b>	<b>32.1 ± 0.6</b>
Other aromatics	<b>3.2 ± 0.6</b>	<b>5.8 ± 1.1</b>	<b>4.4 ± 0.7</b>	<b>5.3 ± 1.0</b>	<b>2.6 ± 0.5</b>	<b>5.5 ± 1.1</b>	<b>6.6 ± 1.3</b>	<b>4.8 ± 1.0</b>	<b>6.2 ± 1.1</b>
Sesquiterpenoids	<b>0.5 ± 0.2</b>	-	<b>1.6 ± 0.3</b>	<b>0.6 ± 0.2</b>	<b>13.3 ± 2.3</b>	<b>9.5 ± 1.7</b>	<b>3.5 ± 0.6</b>	-	<b>0.3 ± 0.1</b>

Table II (continued)

Group of compounds	Polish, Pr-1	Polish, Pr-2	Russian, Pr-3	Slovak, Pr-4	Russian, Pr-5	Russian, Pr-6	Russian, Pr-7	Finnish, Pr-8	Latvian, Pr-9
Triterpenoids	1.9 ± 0.4	2.2 ± 0.5	1.8 ± 0.5	3.5 ± 0.7	0.3 ± 0.1	Trace	8.8 ± 1.6	3.4 ± 1.1	2.7 ± 0.5
Aliphatic acids and esters, including Tetracosyl and hexacosyl palmitates	6.3 ± 1.4	19.6 ± 2.9	20.4 ± 1.2	13.8 ± 0.8	7.9 ± 1.4	13.0 ± 0.9	18.2 ± 3.4	7.8 ± 1.5	5.8 ± 1.2
Aliphatic alcohols	1.0 ± 0.3	8.2 ± 1.6	8.8 ± 1.7	5.6 ± 1.4	1.1 ± 0.3	2.4 ± 0.5	7.6 ± 1.7	2.1 ± 0.6	1.9 ± 0.5
Alkanes and alkenes	0.5 ± 0.3	0.5 ± 0.4	1.3 ± 0.3	0.6 ± 0.2	0.2 ± 0.01	0.1 ± 0.1	1.9 ± 0.3	–	–
Nonidentified	7.7 ± 0.5	18.4 ± 0.4	18.8 ± 0.4	10.7 ± 0.3	7.1 ± 0.2	15.2 ± 0.3	11.0 ± 0.7	6.4 ± 0.4	6.5 ± 0.4
	2.7 ± 0.3	2.8 ± 0.3	2.4 ± 0.4	2.4 ± 0.5	13.1 ± 2.4	7.1 ± 0.9	7.3 ± 1.0	4.6 ± 1.1	9.8 ± 2.0

Hierarchical cluster analysis was performed based on chemical data (Table II). As shown in Figure 2, nine propolis samples can be separated into three groups. Samples Pr-2, Pr-3, Pr-4 form one group harvested from poplar buds. Pure poplar-type sample Pr-1 belong to the same group, showing a difference from other samples of this group with small admixtures (0.6–1.0%) of aspen bud resins. Samples Pr-5, Pr-6, and Pr-7 were grouped as another group harvested presumably from birch buds. A slight difference shows sample Pr-5 which contains the least admixture of aspen resins. The third group was formed of samples Pr-8 and Pr-9 harvested from aspen buds, which are considerably different from other samples in anti-*P. larvae* activity (see below).

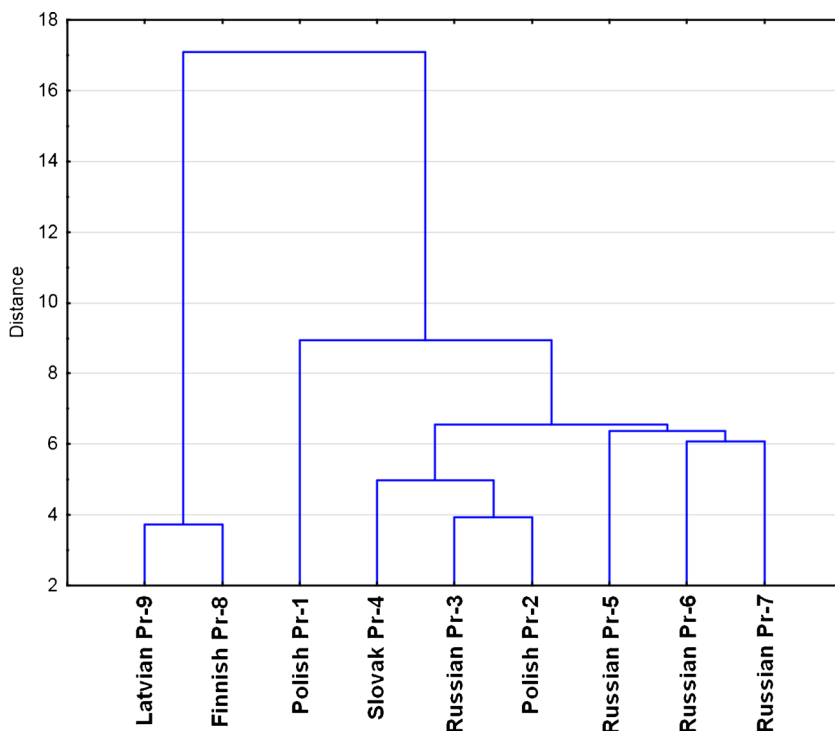
To test the antimicrobial activity of the propolis extracts, we used the tube dilution assay. The sensitivity of the “wild” isolates and reference LMG 09820 strain of *P. larvae* were tested. The minimal inhibitory concentrations of the propolis extracts are presented in Table III. The most active was the Polish poplar-type propolis Pr-1, which showed MIC values that ranged from 7.8 to 15.6  $\mu\text{g mL}^{-1}$ . In contrast, the Latvian pure aspen-type propolis Pr-9 showed lower antimicrobial activity (from 31.8 to 62.5  $\mu\text{g mL}^{-1}$ ) against all tested strains.

#### 4. DISCUSSION

It is relevant to ascertain how the chemical composition of different propolis types influences their activity against *P. larvae*. As can be observed in Table III, despite qualitative and quantitative differences in their composition, all tested extracts significantly inhibited the growth of the Polish isolates and the reference strain of *P. larvae* with MIC values ranging from 7.8 to 62.4  $\mu\text{g mL}^{-1}$ .

The investigated poplar-type propolis (P-1 to P-4) showed practically identical effects on the pathogen, which is not surprising considering that their qualitative composition was very similar. Moreover, the quantitative compositions of the biologically active phenols in P-1 to P-4 were similar if the lipophilic wax components were excluded. Therefore, the slightly higher activity of Pr-1 with a MIC value of 7.8  $\mu\text{g mL}^{-1}$  in the assay with KB25, KB35, and KB41 isolates might be due to the lower (compared to propolis samples





**Figure. 2** Hierarchical cluster analysis of propolis samples based on their chemical composition.

Pr-2 to Pr-4) relative content of “neutral” wax components such as saturated fatty acids and their esters with saturated aliphatic alcohols (6.3% of

TIC), alkanes, and alkenes (7.7% of TIC), which do not influence the biological activity of propolis. Notably, samples Pr-1 and Pr-2 were collected

**Table III.** Minimal inhibitory concentration of different extracts for *P. larvae*.

Propolis	MIC ( $\mu\text{g mL}^{-1}$ ) for <i>P. larvae</i> strains				
	KB25	KB35	KB41	KB55	LMG 09820
<b>Poplar type</b>					
Pr-1 (Poland-1)	7.8	7.8	7.8	15.6	15.6
Pr-2 (Poland-2)	15.6	15.6	15.6	15.6	15.6
Pr-3 (Russia-1)	15.6	15.6	15.6	15.6	15.6
Pr-4 (Slovak)	15.6	15.6	15.6	15.6	15.6
<b>Birch type</b>					
Pr-5 (Russia-2)	15.6	15.6	15.6	15.6	15.6
Pr-6 (Russia-3)	15.6	31.8	15.6	15.6	15.6
Pr-7 (Russia-4)	15.6	31.8	15.6	31.8	31.8
<b>Aspen type</b>					
Pr-8 (Finland)	31.8	31.8	15.6	15.6	31.8
Pr-9 (Latvia)	31.8	62.4	31.8	62.4	62.4

at the same time by honeybees inhabiting neighboring hives. It appears that one of the two bee colonies, for unknown reasons, added more waxes to the collected poplar bud resins.

The high anti-*P. larvae* activity of the poplar-type and birch-type propolis can be attributed to the high content of flavonoids, which is consistent with recent findings (Popova et al. 2007; Mihai et al. 2012; Bilikova et al. 2013). However, a substantial difference in the composition of the flavonoid fraction of these two types of propolis was notable. The products derived from poplar bud resin contained large amounts of dihydroflavonols (pinobanksin and its 3-substituted derivatives) and the flavanones pinocembrin and pinostrobin, which were completely absent in birch-type propolis (Table II). On the contrary, only small amounts of the main birch-bud flavanone sakuranetin was present in poplar-type propolis. Sakuranetin and another flavanone, homoeriodictyol, as well as the flavone pectolinarigenin can be considered as taxonomic markers of downy birch (*B. pubescens*) on par with phenylpropenoid sesquiterpenols (Isidorov et al. 2014b, 2016). These flavonoids possess antimicrobial activity (Atkinson and Blakeman 1982; Grecco et al. 2014). Therefore, they may contribute to the spectrum of biological activity of propolis from the boreal zone of Europe, including anti-*P. larvae* action.

Although they only contained trace amounts of flavonoids, the aspen-type samples of propolis were active against *P. larvae* (Table III). This activity can be attributed to other phenols, i.e., hydroxycinnamyl glycerides. This hypothesis needs experimental confirmation; phenylpropenoid glycerides have demonstrated antitumor, antiproliferative, and estrogenic activity (Banskota et al. 2002; Cheng et al. 2007; Gunasekera et al. 1981), although their antimicrobial properties have not been previously investigated. Notably, aspen-type Pr-8 propolis, containing an admixture of birch bud-derived phenylpropenoid sesquiterpenols, was slightly more active than the pure aspen-type sample Pr-9.

These findings firstly assume that not only flavonoids but also different phenylpropenoids (such as phenylpropenoid glycerides, the main phenolics of aspen-type propolis) are also responsible for the antibacterial activity of propolis from the boreal zone. Secondly, the higher anti-*P. larvae* activity

of Pr-8 propolis, which was “polluted” with components derived from birch bud resin, might provide evidence of a synergistic interaction among different compounds in propolis (Mihai et al. 2012). Resin diversity may be beneficial for bees with respect to protection against different pathogens (Drescher et al. 2014; Wilson et al. 2013, 2015). It is relevant and worthwhile to further study the interaction effects among phenols of different classes of propolis extracts.

Finally, a comparison of the results of investigations dealing with the in vitro anti-*P. larvae* action of essential oils and propolis confirmed the higher activity of propolis. For example, essential oils from four *Citrus* species demonstrated MIC values ranging from 385 to 840  $\mu\text{g mL}^{-1}$  (Fuselli et al. 2008). According to Gende et al. (2008, 2009), relatively high MIC values (from 50 to 300  $\mu\text{g mL}^{-1}$ ) also characterize essential oils from *Lavandula officinalis*, *Cinnamomum zeylanicum*, *Mentha piperita*, *Pimpinella anisum*, and *Foeniculum vulgare*. Therefore, the application of propolis for nonmedicinal AFB treatment appears to be more promising.

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**Contributions** VI—propolis collection and GC-MS analysis; KB—*P. larvae* isolation and identification; GZ and IS—microbiological investigations; KM—participation in the design of the data

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**Etude in vitro de l'activité anti-microbienne de propolis d'origine européenne vis-à-vis de *Paenibacillus larvae***

*Paenibacillus larvae* / loque américaine / *Apis mellifera* / propolis / action antibactérienne

**In vitro Studie zur antimikrobiellen Aktivität von europäischem Propolis gegenüber *Paenibacillus larvae***

***Paenibacillus larvae* / Amerikanische Faulbrut / *Apis mellifera* / Propolis / antibakterielle Aktivität**

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